

An Infinite Family of Networks with Multiple Non-Degenerate Equilibria

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What are Chemical Reaction Networks?

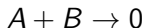
Definition

A chemical reaction network $G = \{\mathcal{S}, \mathcal{C}, \mathcal{R}\}$ consists of three finite sets:

- 1 a set of species \mathcal{S} ,
- 2 a set \mathcal{C} of complexes, and
- 3 a set $\mathcal{R} \subseteq \mathcal{C} \times \mathcal{C}$ of reactions

Example

Let $\mathcal{S} = \{A, B\}$, $\mathcal{C} = \{0, A, B, A + B\}$ and $\mathcal{R} = \{(A + B, 0), (A, B)\}$ then, the network associated is:



The Reaction Kinetics System

Definition

The *stoichiometric matrix* Γ is the $|\mathcal{S}| \times |\mathcal{R}|$ matrix whose k^{th} column is the vector $y_j - y_i$ of the k^{th} reaction $y_i \rightarrow y_j$.

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Definition

The *reactant vector* $\rho(x) \in \mathbb{R}^{|\mathcal{R}|}$ is the vector whose k^{th} entry represents the k^{th} reactant complex:

$$\alpha_1 X_1 + \alpha_2 X_2 + \cdots + \alpha_{|\mathcal{S}|} X_{|\mathcal{S}|}$$

as the product:

$$r_k X_1^{\alpha_1} X_2^{\alpha_2} \cdots X_{|\mathcal{S}|}^{\alpha_{|\mathcal{S}|}}$$

where $r_k \in \mathbb{R}^+$ is the reaction rate of the k^{th} reaction.

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The **reaction kinetics system** defined by a reaction network G is given by the following system of ODEs:

$$\frac{dx}{dt} = \Gamma \cdot \rho(x)$$

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Definition

A **steady state** of a reaction kinetics system $\frac{dx}{dt} = \Gamma \cdot \rho(x)$ is a non-negative concentration vector $x^* \in \mathbb{R}_{\geq 0}^{|\mathcal{S}|}$ for which $\Gamma \cdot \rho(x^*) = 0$.

Definition

A steady state $x^* \in \mathbb{R}_{> 0}^{|\mathcal{S}|}$ is **nondegenerate** if $\text{im}(df(x^*)) = \text{im}(\Gamma)$, where $df(x^*)$ denotes the Jacobian of the reaction kinetics system at x^* .

A Good Candidate: The Network $\tilde{K}_{m,n}$

For positive integers $n \geq 2$, $m \geq 1$ we define the network $\tilde{K}_{m,n}$ of order n and production factor m to be:

$$X_1 + X_2 \rightarrow 0$$

$$\vdots$$

$$X_{n-1} + X_n \rightarrow 0$$

$$X_1 \rightarrow mX_n$$

$$X_j \rightarrow 0$$

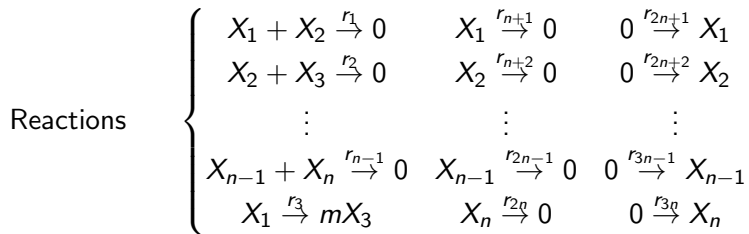
$$0 \rightarrow X_j$$

Theorem (Shiu & Joshi, 2015)

For positive integers $n \geq 2$ and $m \geq 2$, the fully open extension $\tilde{K}_{m,n}$ is multistationary if n is odd.

Some Properties of $\tilde{K}_{m,n}$

For any n , the system is given by,



$$\text{ODE's} \left\{ \begin{array}{l} \dot{x}_1 = -r_1 x_1 x_2 - r_n x_1 - r_{n+1} x_1 + r_{2n+1} \\ \dot{x}_i = -r_{i-1} x_{i-1} x_i - r_i x_i x_{i+1} - r_{n+i} x_i + r_{2n+i}, \text{ for } 2 \leq i \leq n-1 \\ \dot{x}_n = -r_{n-1} x_{n-1} x_n + m r_n x_1 - r_{2n} x_n + r_{3n} \end{array} \right.$$

Jacobian Matrix of $\tilde{K}_{m,n}$

$$df(\mathbf{x})_{(1,1)} = -r_1 x_2 - r_n - r_{n+1}$$

$$df(\mathbf{x})_{(1,2)} = -r_1 x_1$$

$$df(\mathbf{x})_{(i,i-1)} = -r_{i-1} x_i \quad \forall i \in \{2, 3, \dots, n-1\}$$

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$$df(\mathbf{x})_{(n,n)} = -r_{n-1} x_{n-1} - r_{2n}$$

Jacobian Matrix of $\tilde{K}_{m,n}$

$$\begin{bmatrix}
 -r_1 x_2 - r_n - r_{n+1} & -r_1 x_1 & 0 & \dots & 0 & 0 \\
 -r_1 x_2 & -r_1 x_1 - r_2 x_3 - r_{n+2} & -r_2 x_2 & \dots & \vdots & \vdots \\
 0 & -r_2 x_3 & -r_2 x_2 - r_3 x_4 - r_{n+3} & \ddots & 0 & 0 \\
 0 & 0 & -r_3 x_4 & \ddots & -r_{n-2} x_{n-2} & 0 \\
 \vdots & \vdots & \vdots & \ddots & -r_{n-2} x_{n-2} - r_{n-1} x_n - r_{2n-1} & -r_{n-1} x_{n-1} \\
 mr_n & 0 & 0 & \dots & -r_{n-1} x_n & -r_{n-1} x_{n-1} - r_{2n}
 \end{bmatrix}$$

Goal

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Find rates r_i and two steady state concentrations, \mathbf{x}^* , $\mathbf{x}^\#$, and show $Im(df(\mathbf{x})) = Im(\Gamma)$.

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But,

$$\Gamma = \left[\begin{array}{cccccc|c|c} -1 & 0 & 0 & \dots & 0 & -1 & & \\ -1 & -1 & 0 & \dots & 0 & 0 & & \\ 0 & -1 & -1 & \dots & \vdots & \vdots & -I^n & I^n \\ 0 & 0 & -1 & \ddots & \vdots & \vdots & & \\ \vdots & \vdots & \vdots & \ddots & -1 & 0 & & \\ 0 & 0 & 0 & \dots & -1 & m & & \end{array} \right]$$

Hence, $Im(\Gamma) = \mathbb{R}^n$

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Hence, $\text{Im}(\Gamma) = \mathbb{R}^n$

NEW GOAL:

Find rates r_i and two steady state concentrations, \mathbf{x}^* , $\mathbf{x}^\#$, and show $\det(df(\mathbf{x})) \neq 0$.

The problem

Conjecture (Shiu & Joshi, 2015)

For positive integers $n \geq 2$ and $m \geq 2$, if n is odd, then $\tilde{K}_{m,n}$ admits multiple positive **non-degenerate** steady states.

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Approach

Backtrack the **Determinant Optimization Method** by Craicun & Feinberg, 2005

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If two conditions hold for a chemical reaction system, then it has the capacity for at least two steady state equilibria.

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The main conditions on internal and outflow reactions:

$$(I) \quad \det(y_1, y_2, \dots, y_n) \cdot \det((y_1 - y'_1), (y_2 - y'_2), \dots, (y_n - y'_n)) < 0$$

$$(II) \quad \sum_{i=1}^k \tilde{\eta}_i (y_i - y'_i) \in \mathbb{R}_+^S \text{ for positive numbers } \tilde{\eta}_1, \dots, \tilde{\eta}_k.$$

Remark: The network $\tilde{K}_{m,n}$ satisfies **both** conditions.

Define the **augmented Jacobian**,

$$T_\eta = \begin{bmatrix} \eta_1 + \eta_n + \eta_{n+1} & \eta_1 & 0 & \cdots & 0 \\ \eta_1 & \eta_1 + \eta_2 + \eta_{n+2} & \eta_2 & \cdots & 0 \\ 0 & \eta_2 & \ddots & & \cdots 0 \\ \vdots & 0 & \ddots & & \ddots \\ 0 & \vdots & \cdots & \eta_{n-2} + \eta_{n-1} + \eta_{2n-1} & \eta_{n-1} \\ -m \eta_3 & 0 & \cdots & \eta_{n-1} & \eta_{n-1} + \eta_{2n} \end{bmatrix}.$$

We define, for large λ and small ϵ , η_i to be:

$$\begin{cases} \lambda & \text{for } 1 \leq i \leq n-2 \text{ and } i = n \\ (m+1)\lambda & \text{for } i = n-1 \\ \epsilon & \text{for } n+1 \leq i \leq 2n \end{cases}$$

Delta Vector

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We use this to find $\delta \in \mathbb{R}_{\neq 0}^{|S|}$ such that $T_{\eta^0} \cdot \delta = 0$.

Letting $\eta^0 = \eta^-$ for all entries excluding η_{2n}^0 , we see

$$T_{\eta^0} = \begin{bmatrix} 2\lambda + \epsilon & \lambda & 0 & \dots & 0 \\ \lambda & 2\lambda + \epsilon & \lambda & \dots & 0 \\ 0 & \lambda & 2\lambda + \epsilon & \lambda \dots & 0 \\ \vdots & 0 & \ddots & \ddots & \ddots \\ 0 & \vdots & \dots & \lambda + \lambda(m+1) + \epsilon & \lambda(m+1) \\ -m\lambda & 0 & \dots & \lambda(m+1) & \lambda(m+1) + \eta_{2n}^0 \end{bmatrix}.$$

The general solution of η_{2n}^0

$$\eta_{2n}^0 = \frac{(m+1)(m\lambda^n + \lambda(m+1)\Upsilon_{n-2})}{(\lambda(m+2) + \epsilon)\Upsilon_{n-2} - \lambda^2\Upsilon_{n-3}} - \lambda(m+1)$$

The general solution of η_{2n}^0

$$\eta_{2n}^0 = \frac{(m+1)(m\lambda^n + \lambda(m+1)\tau_{n-2})}{(\lambda(m+2) + \epsilon)\tau_{n-2} - \lambda^2\tau_{n-3}} - \lambda(m+1)$$

where

$$\tau_i = \frac{1}{2^{i+1}(\epsilon)^{\frac{1}{2}}(\epsilon + 4\lambda)^{\frac{1}{2}}} \cdot (-\epsilon + 2\lambda(\epsilon + 2\lambda - (\epsilon)^{\frac{1}{2}}(\epsilon + 4\lambda)^{\frac{1}{2}}))^i +$$
$$(\epsilon)^{\frac{1}{2}}(\epsilon + 4\lambda)^{\frac{1}{2}}(\epsilon + 2\lambda - (\epsilon)^{\frac{1}{2}}(\epsilon + 4\lambda)^{\frac{1}{2}}))^i + \epsilon + 2\lambda((\epsilon)^{\frac{1}{2}}$$
$$(\epsilon + 4\lambda)^{\frac{1}{2}} + \epsilon + 2\lambda))^i + (\epsilon)^{\frac{1}{2}}(\epsilon + 4\lambda)^{\frac{1}{2}}((\epsilon)^{\frac{1}{2}}$$
$$(\epsilon + 4\lambda)^{\frac{1}{2}} + \epsilon + 2\lambda))^i$$

Delta Recurrence

Next, we find $\delta \in \mathbb{R}_{\neq 0}^{|S|}$ such that $T_{\eta^0} \cdot \delta = 0$. Note that the nullspace of T_{η^0} is non trivial, since $\det(T_{\eta^0}) = 0$. We let

$$\delta_0 = 0 \quad (\text{For convenience})$$

$$\delta_1 = \delta_1$$

$$\delta_k = \frac{-(2\lambda + \epsilon)}{\lambda} \delta_{k-1} - \delta_{k-2} \text{ for } 2 \leq k \leq n-1$$

$$\delta_n = \frac{-(\lambda(m+2) + \epsilon)}{\lambda(m+1)} \delta_{n-1} - \frac{1}{m+1} \delta_{n-2}$$

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The recurrence is given by

$$\delta_k = \delta_1 \lambda \cdot \frac{(\sqrt{4\lambda\epsilon + \epsilon^2} - (2\lambda + \epsilon))^k - (-\sqrt{4\lambda\epsilon + \epsilon^2} - (2\lambda + \epsilon))^k}{2^k \lambda^k \sqrt{4\lambda\epsilon + \epsilon^2}}$$

Concentrations & Rates

- Now we use $\delta \in NS(T_{\eta^0})$, to define all rates

$$r_i = \frac{\langle y_i, \delta \rangle}{e^{\langle y_i, \delta \rangle} - 1} \eta_i^0$$

and concentrations

$$\mathbf{x}^* = (1, 1, \dots, 1)$$

$$\mathbf{x}^\# = (e^{\delta_1}, e^{\delta_2}, \dots, e^{\delta_n}),$$

which are proven to be TWO distinct steady states.

Example $n = 3$

Using the formulas in previous slides, we compute our rates:

$$r_1 = \frac{-1.1}{e^{-1.1}-1} \approx 1.65 \quad r_2 = \frac{1.31}{e^{\frac{1.31}{m+1}} - 1} \quad r_3 = \frac{1}{e-1} \approx .58$$

$$r_4 = \frac{.1}{e-1} \approx .06 \quad r_5 = \frac{-.21}{e^{-2.1}-1} \approx .24 \quad r_6 = \frac{m - 1.31}{e^{\frac{2.1m+3.41}{m+1}} - 1}$$

and concentrations:

$$\mathbf{x}^* = (1, 1, 1)$$
$$\mathbf{x}^\# = (e, e^{-2.1}, e^{\frac{2.1m+3.41}{m+1}})$$

Note that only $\mathbf{x}_3^\#$, r_2 and r_6 depend on m .

The Determinant of Jacobians

By substitution we obtain the determinant of the Jacobian for the system:

$$\det(df(\mathbf{x}^*)) = r_2 r_1 r_3 m - (r_2 + r_6)(r_1 r_3 + r_1 r_4 + r_1 r_5 + r_3 r_5 + r_4 r_5) - r_2 r_6 (r_1 + r_3 + r_4)$$

$$\det(df(\mathbf{x}^\#)) = r_2 x_2 ((r_1 x_2 + r_3 + r_4)(r_2 x_3) + r_1 x_1 m r_3) - (r_2 x_2 + r_6)(r_1 x_2 + r_3 + r_4)(r_1 x_1 + r_2 x_3 + r_5) + (r_2 x_2 + r_6)(r_1 x_1 r_1 x_2)$$

Bounding the determinants

Based on the following inequalities,

$$0.14m > r_6 = \frac{m - 1.31}{e^{\frac{2.1m+3.41}{m+1}} - 1} > 0.13m - 0.5$$

$$m + 1 > r_2 = \frac{1.31}{e^{\frac{1.31}{m+1}} - 1} \geq m$$

$$e^{\frac{2.1y+3.41}{y+1}} > x_3 = e^{\frac{2.1m+3.41}{m+1}} > e^{2.1} \quad \forall m \geq y$$

we can bound the determinants of our Jacobians,

Proven Bounds

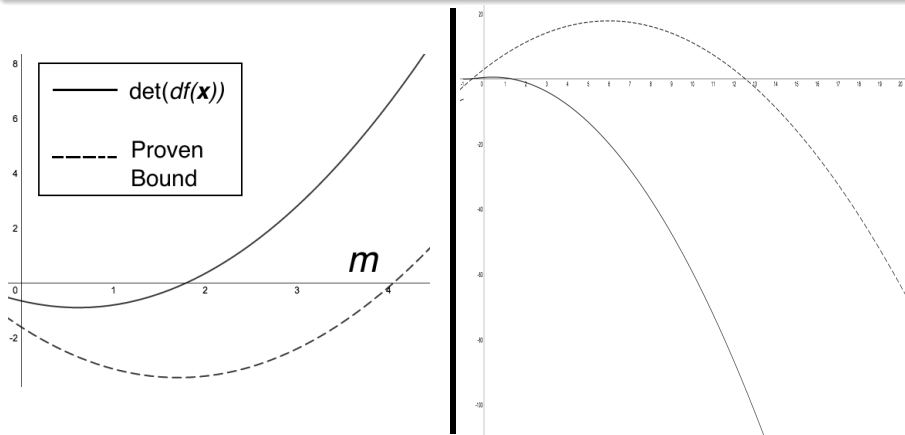
$$\det(df(\mathbf{x}^*)) > 0.6294m^2 - 2.156m - 1.61 \quad \forall m \geq 2$$

$$\det(df(\mathbf{x}^\#)) < -0.41295m^2 + 4.9437m + 3.06205. \quad \forall m \geq 20$$

Bounding the determinants

Theorem

The chemical reaction system $\tilde{K}_{m,3}$ has multiple positive non-degenerate steady states for $m \geq 2$.



Conclusion

- 1 We have partially solved the conjecture for $n = 3$ and any m .
- 2 We have also solved the conjecture for $n \leq 11$ and small m by using our approach. We have showed that the Determinant Optimization Method can create **degenerate** steady states

Some future directions:

- 1 A more interesting problem is to solve the conjecture for fixed m , in particular, $m = 2$.
- 2 Find an alternate method to get closed forms for the steady states of a chemical reaction network.
- 3 Find different criteria to characterize steady states as non degenerate.

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